



At the heart of every scientific endeavor is a laboratory where the research experiments and analytical procedures are accomplished. For over a hundred years, laboratories have counted on Praxair to fulfill their critical gas needs. Praxair offers a comprehensive range of gas products with the proper purity, compliance, and certification to meet the demands of any laboratory application. Praxair also offers the industry's leading collection of gas handling, distribution, and storage equipment to help lab personnel work safely and productively. Praxair's focus on quality and service gives laboratories the confidence to know that the only variables they need to worry about are in their experiments, not their gases.

Comprehensive Product Range

Supplying each application with the proper gas (grade, compliance, certification, scale).

Superior Reliability

Ensuring high quality products are there when they are needed.

Equipment and Systems Excellence

Connecting customers with everything needed to use, handle, and store gases efficiently and safely.

Productivity and Innovation Partner

Working with laboratories to identify and implement productivity and cost improvements.

Ease of Doing Business

Providing convenient e-commerce options and strong customer support.

Safety Focus

Making safety the top priority for every activity.

Equipment and Systems Excellence

Praxair manufactures and supplies thousands of gas handling, storage, and safety solutions:

- regulators
- flow meters
- gas distribution systems
- gas purification accessories
- gas generators and back-up systems
- freezers and cryogenic systems
- valves, fittings, and CGA connections
- gas cabinets
- gas detection and safety equipment

...and much more.

Praxair designs and fabricates all gas distribution systems in-house using high quality components. For laboratories pursuing renovation or expansion, Praxair can design and implement a turn-key build-out.

Protect the purity of your gases and the quality of your lab's results with the right product for the right application.



See Section E for additional information on Praxair's full line of ProSpec™ gas handling solutions



Multi-Gas Point-of-Use-Panel, see page E•303



5029 Protocol Station, see page E•281



4092 Regulator, see page E•266



Dual Cylinder Rack with Automatic Switch-Over, see page E•297

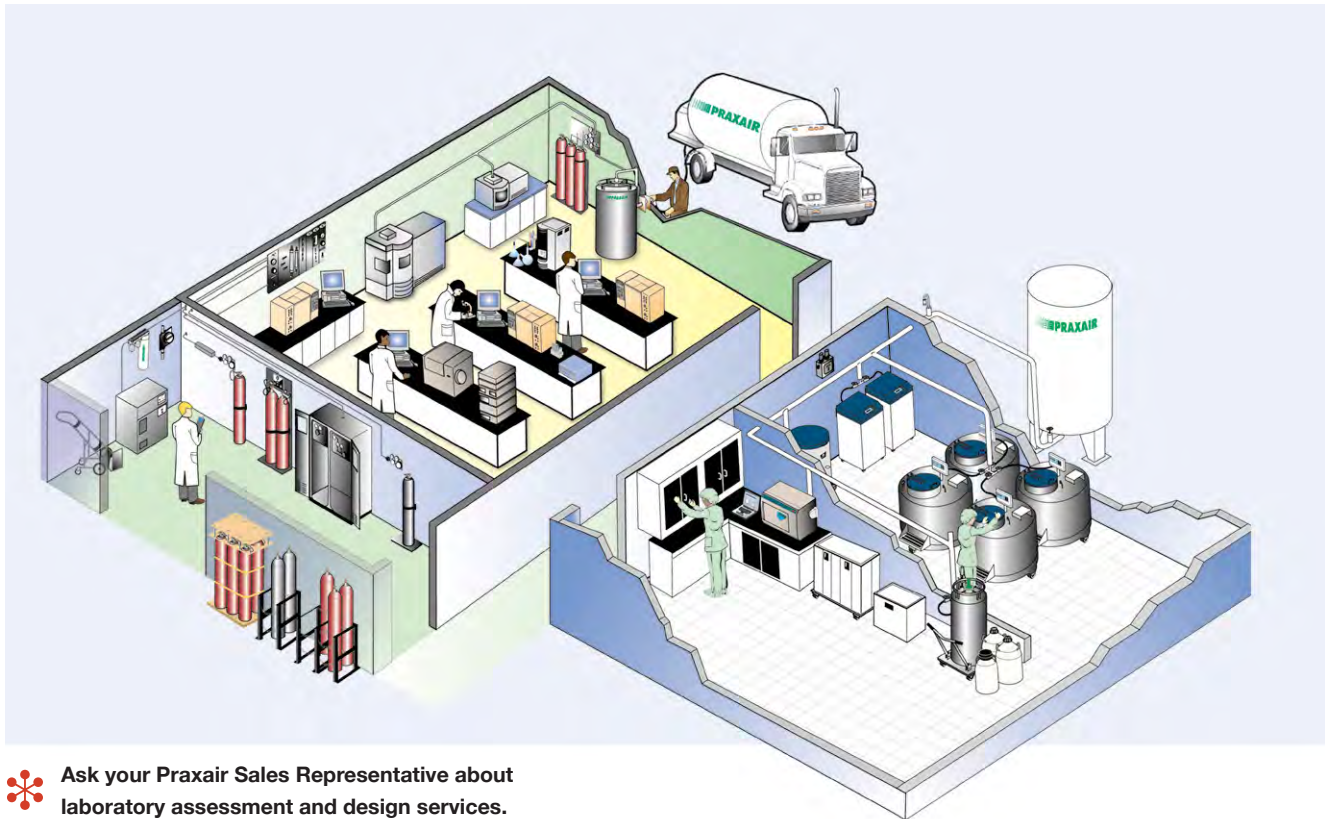
Laboratory Assessment Service


Praxair applies its extensive experience with specialty gases and equipment to help laboratory managers and scientists review their gas needs and select the proper gas products and delivery systems. Praxair's Laboratory Assessment Program examines all aspects of gas use, handling, and storage to help laboratories reduce costs and mitigate risk.



Laboratory Assessment Benefits

- Achieve consistent work quality
- Improve productivity
- Optimize supply modes
- Protect instrumentation and equipment
- Better utilize bench and floor space
- Implement appropriate safety measures
- Streamline transaction processing
- Ensure appropriate product grades and certifications
- Improve asset management



 **Ask your Praxair Sales Representative about laboratory assessment and design services.**
We're here to help!

Praxair's Gas Grade Recommendations

Analytical Method/ Detector	Instrument Gases	Page Number	Analytical Range		Type of Analysis Impurity Considerations
			< 100 ppm	> 100 ppm	
Gas Chromatography					
TCD (Thermal Conductivity Detector)	Ar	B•26	AR 5.5TG	AR 5.0UH	Universal Detector Atmospheric contaminants can oxidize the detector filament giving rise to negative peaks and reduced sensitivity.
	He	B•53	HE 5.5TG	HE 5.0UH	
	H ₂	B•54	HY 6.0RS	HY 5.0UH	
	N ₂	B•65	NI 5.5TG	NI 5.0UH	
FID (Flame Ionization Detector)	Air	B•24	AI 0.0UZ	AI 0.0UZ	Organic Compounds Hydrocarbons in carrier and fuel gases can give rise to baseline noise and reduced detector sensitivity. Oxygen and water cause column deterioration and affect retention time on critical separations.
	Ar	B•26	AR 5.5TG	AR 5.0UH/ AR 4.8Z	
	He	B•53	HE 5.5TG	HE 5.0UH/ HE 4.6Z	
	H ₂	B•54	HY 6.0RS	HY 5.0UH/ HY 4.5Z	
	N ₂	B•65	NI 5.5TG	NI 5.0UH/ NI 4.8Z	
	40% H ₂ in He 40% H ₂ in N ₂	D•218 D•218	IG F12UH IG F14UH	IG F11 IG F13	
ECD (Electron Capture Detector)	He	B•53	HE 5.5EC/ HE 5.5TG	HE 5.5TG	Electronegative Functional Groups Detector response and column life are reduced by oxygen and water. Hydrocarbons and halocarbons can produce baseline noise, negative peaks, and plumbing contamination.
	N ₂	B•65	NI 5.5EC/ NI 5.5TG	NI 5.5TG	
	5% CH ₄ in Ar (ECD P-5)	D•218	IG ECD1	IG ECD1	
	10% CH ₄ in Ar (ECD P-10)	D•218	IG ECD2	IG ECD2	
FPD (Flame Photometric Detector)	Air	B•24	AI 0.0UZ	AI 0.0Z	Sulfur or Phosphorous Compounds Organics can yield baseline noise and carbon dioxide can suppress detector response.
	He	B•53	HE 5.5TG	HE 5.0UH	
	H ₂	B•54	HY 6.0RS	HY 5.0UH	
	N ₂	B•65	NI 5.5TG	NI 5.0UH	
PID (Photo Ionization Detector)	He	B•53	HE 5.5TG	N/A	Selective Detector Dependent on UV Source Organics can yield baseline noise and carbon dioxide can suppress detector response.
	H ₂	B•54	HY 6.0RS	N/A	
	N ₂	B•65	NI 5.5TG	N/A	
MS (Mass Spectrometer)	Ar	B•26	AR 5.5TG	AR 5.0UH	Universal Detector Organics can yield baseline noise and carbon dioxide can suppress detector response.
	He	B•53	HE 5.5TG	HE 5.0UH	
	H ₂	B•54	HY 6.0RS	HY 5.0UH	
	N ₂	B•65	NI 5.5TG	NI 5.0UH	
DID (Discharge Ionization Detector)	He	B•53	HE 6.0RS	N/A	Universal Detector Atmospheric impurities can cause baseline noise signal polarity and reduced detector stability and sensitivity.
	He Purge	B•53	HE 5.0UH	N/A	
USD (Ultrasonic Detector)	Ar	B•26	AR 5.0UH	AR 5.0UH/ IAR 4.8Z	Universal Detector Atmospheric impurities can cause baseline noise signal polarity and reduced detector stability and sensitivity.
	He	B•53	HE 5.0UH	HE 5.0UH/ HE 4.6Z	

Praxair's Gas Grade Recommendations

Analytical Method Detector	Instrument Gases	Page Number	Analytical Range		Type of Analysis Impurity Considerations
			< 100 ppm	> 100 ppm	
Optical Spectrometry – Absorption					
IR (Infrared)	Air	B•24	AI 0.0UZ	AI 0.0Z	Polyatomic and Heteroatomic Compounds Oxygen can oxidize a sample. Moisture interferes with IR spectra. Impurities coinciding with analyte peaks can cause inaccuracies.
	Ar	B•26	AR 5.0UH	AR 5.0UH	
	N ₂	B•65	NI 5.0UH	NI 5.0UH	
	LN ₂	B•85	Liquid	Liquid	
AA (Atomic Absorption)	Air	B•24	AI 0.0Z	AI 0.0Z	Elemental Analysis Impurities can cause the flame to discolor or burn unevenly. Furnace atmospheres require low oxygen and moisture levels to maintain instrument sensitivity.
	Ar	B•26	AR 5.0UH	AR 5.0UH	
	C ₂ H ₂	B•23	AC 2.6AA	AC 2.6AA	
	n-C ₄ H ₁₀	B•31	BU 2.5IS	BU 2.5IS	
	H ₂	B•54	HY 4.5Z	HY 4.5Z	
	N ₂	B•65	NI 5.0UH	NI 5.0UH	
	N ₂ O	B•69	NS 2.5AA	NS 2.5AA	

Gas Grade Legend

- AA** – Atomic Absorption
- EC** – Electron Capture Detector
- IS** – Instrument
- RS** – Research/Chromatographic
- TG** – Trace Analytical
- UH** – Ultra High Purity
- UZ** – Ultra Zero
- Z** – Zero

Please refer to designated page numbers for available cylinder styles, contents, and other information.

Please refer to the Instrumentation Mixture Summary on page D•218 for additional information regarding mixtures.

Please refer to the Regulator Reference Guide on page E•241 for help in selecting the appropriate cylinder regulator.

Application Note: Helium and Hydrogen Carrier Gases for Gas Chromatography



Helium and hydrogen are the most common carrier gases used for gas chromatography applications. These two gases deliver similar optimum achievable separation efficiencies (theoretical plates). Hydrogen offers advantages versus helium in having a higher and broader range of linear velocities for which near-optimum separation efficiency can be achieved, especially for applications requiring longer columns (higher inlet pressures). The higher linear velocities available to hydrogen enable significant potential reductions in analysis time. Unlike inert helium, hydrogen has the potential to react with some analytes; such reactions are uncommon under normal gas chromatography conditions. Users should note that conversion from one carrier gas to

another can involve significant changes to standard operating procedures. In executing a conversion from helium to hydrogen, one must consider not only compatibility of the gas chromatograph but also the associated plumbing and connected components (e.g., filters or mass spectrometer). Hydrogen presents flammability and asphyxiation hazards, and appropriate safety precautions should be taken in accordance with careful risk assessment when working with hydrogen in a lab. Helium is typically supplied via cylinders, while hydrogen can be supplied via cylinders or on-site generators. With decades of experience in supplying gases for analytical instruments, Praxair is prepared to help users identify and move forward their preferred best gas solutions for their particular applications.

Praxair's Gas Grade Recommendations

Analytical Method Detector	Instrument Gases	Page Number	Analytical Range		Type of Analysis Impurity Considerations
			< 100 ppm	> 100 ppm	
Optical Spectrometry – Atomic Emission					
ICP (Inductively Coupled Plasma)	Ar	B•26	AR 4.8IC	AR 4.8IC	Elemental Analysis (see application note below)
	LAr	B•82	AR 4.8IC	AR 4.8IC	
	N ₂	B•65	NI 5.0UH	NI 5.0UH	
Flame	Air	B•24	AI 0.0Z	AI 0.0Z	Elemental Analysis
	C ₂ H ₂	B•23	AC 2.6AA	AC 2.6AA	
	n-C ₄ H ₁₀	B•31	BU 2.5IS	BU 2.5IS	
	H ₂	B•54	HY 4.5Z	HY 4.5Z	
	N ₂ O	B•69	NS 2.5AA	NS 2.5AA	
Arc or Spark	Ar	B•26	AR 5.0UH	AR 4.8	Elemental Analysis
	H ₂	B•54	HY 5.0UH	HY 4.5Z	
	5% Ar in H ₂	D•218	IG NC6	IG NC6	
Optical Spectrometry – Chemiluminescence					
Chemiluminescence	Air	B•24	AI 0.0CE	AI 0.0CE	Atmospheric Pollutants – NO_x, SO_x, O₃
	N ₂	B•65	NI 5.5CE	NI 5.5CE	
	O ₂	B•70	OX 4.3UH	OX 4.3UH	
Optical Spectrometry – Fluorescence					
XRF (X-Ray Fluorescence)	10% CH ₄ in Ar	D•218	IG P10	IG P10	Elemental Analysis
	1.3% n-C ₄ H ₁₀ in He	D•218	IG NC4	IG NC4	
	0.95% i-C ₄ H ₁₀ in He	D•218	IG NC5	IG NC5	
	LN ₂	B•85	NI 4.8LC	NI 4.8LC	
UVF (Ultra-Violet Fluorescence)	Air	B•24	AI 0.0VC	AI 0.0VC	SO₂ - H₂S - Organic Compounds
	N ₂	B•65	NI 5.5CE	NI 5.5CE	

Application Note: Using Argon with ICP



Inductively coupled plasma (ICP) is a popular means of preparing analytes for atomic emission spectrometry (ICP-AES) and mass spectrometry (ICP-MS). In both methods, the high temperature, ionizing environment of the plasma converts analyte molecules into excited-state atoms and ions. AES methods identify and quantify elements in the analyte on the basis of the electromagnetic radiation emitted by the excited-state species, while MS methods separate and quantify the ions on the basis of their mass-to-charge ratio. Argon is almost always used as the plasma gas for ICP applications. High purity argon is required to sustain a robust plasma, prevent instrument contamination, and avoid interferences with analytes. Praxair offers a variety

of argon grades to meet the requirements of most instruments; the AR4.8IC grade is most commonly used. Argon is purified from atmospheric air and may contain trace levels of krypton, which can interfere with ICP-MS analyses for selenium and strontium. Argon flow rates for ICP often exceed 10 liters per minute, making these applications good candidates for cryogenic liquid supply options. Liquid supply via dewars or microbulk can reduce unit costs and disruptions associated with changing cylinders. Cylinders better preserve product and lab space when ICP instrument usage is infrequent. Praxair can help ICP users choose the right grades and supply options for their needs.

Praxair's Gas Grade Recommendations

Analytical Method Detector	Instrument Gases	Page Number	Analytical Range		"Type of Analysis Impurity Considerations"
			< 100 ppm	> 100 ppm	
Others					
NMR (Nuclear Magnetic Resonance)	LHe	B•84	Liquid	Liquid	Analysis of Molecular Structure
	LN ₂	B•85	Liquid	Liquid	
MS (Mass Spectrometry)	Air	B•24	AI 0.0UZ	AI 0.0UZ	All Compounds via Various Ionization Sources
	Ar	B•26	AR 5.5TG/ AR 5.0UH	AR 5.0UH/ AR 4.8IC	
	LAr	B•82	AR 5.0	AR 5.0/ AR 4.8IC	
	He	B•53	HE 5.5TG	HE 5.0UH	
	H ₂	B•54	HY 6.0RS	HY 5.0UH	
	N ₂	B•65	NI 5.5TG/ NI 5.0UH	NI 5.0UH	
	LN ₂	B•85	NI 5.0	NI 5.0/ NI 4.8	
Nuclear Counter	5% CH ₄ in Ar	D•218	IG P5	IG P5	Radioactivity
	10% CH ₄ in Ar	D•218	IG P10	IG P10	
	1.3% n-C ₄ H ₁₀ in He	D•218	IG NC4	IG NC4	
	0.95% i-C ₄ H ₁₀ in He	D•218	IG NC5	IG NC5	
Hydrometer	Air	B•24	AI 0.0UZ	AI 0.0Z	Moisture in All Gases
	N ₂	B•65	NI 6.0RS	NI 5.0UH	
Paramagnetic Analyzer	N ₂	B•65	N/A	NI 5.0UH	Oxygen in All Gases
	O ₂ in N ₂	C•127	Certified Standard	Certified Standard	
Combustion Analyzer	Ar	B•26	AR 5.0UH	AR 5.0UH	Elemental Analysis – Carbon, Nitrogen, Sulfur
	He	B•53	HE 5.0UH	HE 5.0UH	
	N ₂	B•65	NI 5.0UH	NI 5.0UH	
	O ₂	B•70	OX 5.0RS	OX 5.0RS	

Gas Grade Legend

- CE** – Continuous Emissions Monitoring
- ICP** – Inductively Coupled Plasma
- RS** – Research/Chromatographic
- TG** – Trace Analytical
- UH** – Ultra High Purity
- UZ** – Ultra Zero
- VC** – Volatile Organic Compound Free
- Z** – Zero

Please refer to designated page numbers for available cylinder styles, contents, and other information.

Please refer to the Instrumentation Mixture Summary on page D•218 for additional information regarding mixtures.

Please refer to the Regulator Reference Guide on page E•241 for help in selecting the appropriate cylinder regulator.

Instrument Mixture Summary

Part Number	Product Description	Mixture Application	Cylinder Style	CGA Connection
IG F11-K	40% H ₂ in He (FID Fuel) (THC < 0.5 ppm)	Fuel Gas for GC-FID	K	350
IG F12UH-K	40% H ₂ in He UHP (FID Fuel) (THC < 0.1 ppm)	Fuel Gas for GC-FID	K	350
IG F13-K	40% H ₂ in N ₂ (FID Fuel) (THC < 0.5 ppm)	Fuel Gas for GC-FID	K	350
IG F14UH-K	40% H ₂ in N ₂ UHP (FID Fuel) (THC < 0.1 ppm)	Fuel Gas for GC-FID	K	350
IG ECD1-K	5% CH ₄ in Ar (ECD P-5)	Make-Up Gas for GC-ECD	K	350
IG ECD2-K	10% CH ₄ in Ar (ECD P-10)	Make-Up Gas for GC-ECD	K	350
IG P5-K	5% CH ₄ in Ar (P-5)	Proportional Counters	K	350
IG P10-K	10% CH ₄ in Ar (P-10)	Proportional Counters and X-Ray Fluorescence	K	350
IG NC4-K	1.3% n-C ₄ H ₁₀ in He	Geiger-Muller Counters and X-Ray Fluorescence	K	350
IG NC5-K	0.95% i-C ₄ H ₁₀ in He	Geiger-Muller Counters and X-Ray Fluorescence	K	350
IG NC6-K	5% Ar in H ₂	Spark Emission	K	350

Instrumentation mixtures are supplied as certified standards but can also be produced to meet primary standard or non-certified grade specifications.

Please refer to the Regulator Reference Guide on page E•241 for help in selecting the appropriate cylinder regulator.

